



Course specification
2019- 2020
Faculty of Pharmacy
Mansoura University



First Level

Course Specification Pharmaceutical Organic Chemistry 1

University: Mansoura University (MU)
Faculty: Pharmacy
Department: Pharmaceutical Organic Chemistry
Course title: Pharmaceutical Organic Chemistry 1
Course code: PO111

Program on which the course is given	Pharmaceutical Science
Academic Level	First Level, First semester
Date of course specification	12/12/2019

1. Basic Information: Course data:

Course title:	Pharmaceutical organic Chemistry 1	Code: PO111
Specialization:	Pharmaceutical organic Chemistry	
Prerequisite:	-----	
Teaching Hours:	Lecture: 2	Practical: 1
Number of units: (credit hours)	3	

2. Course Aims:

2.1. Gain an understanding of the basic principles of atomic structures.

2.2. Have a good idea the stereochemistry of the chiral organic compounds

2.3. Enable the student to understand the basics of the chemical reactions.

2.4. Recognize the chemical properties of organic compounds.

3. Intended learning outcomes (ILO_s):

a- Knowledge and understanding

a1	Represent various structural formulas of different organic compounds.
a2	understand some of the chemical reaction mechanism.
a3	Recognize the stereochemistry of the organic molecules through studying the
a4	Know the effect of the intermolecular forces in the physical and chemical properties of the organic compounds.



Course specification
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b- Intellectual skills

b1	Predict the kind of stereochemical representations of the organic molecules.
b2	Think critically about the kind of the reactions and their mechanisms.
b3	Investigate the electronic displacement factors affecting chemical and physical properties of the organic molecules.

c- Professional and practical skills

c1	Design and plain various chiral resolution methods for separation of optically active compounds.
c2	Perform chemical reactions to recognize the presence of some functional groups in the organic compounds.
c3	Record data of the experimental experiments.

d- General and transferable skills

d1	Interpret the sequence of chemical reactions and mechanisms.
d2	Estimate lab safety and chemical toxicity.
d3	Differentiate the several types of stereochemical molecules.
d4	Work effectively in a team.

4. Contents:

Week No	Topics	No. of hours	Lecture credit hours	Practical credit hours
1.	Stereochemistry: Isomerism and Conformational Isomers.	2	2 hours	
2.	Stereochemistry: Configurational Isomers.	2	2 hours	
3.	Stereochemistry: Chirality in Systems lacking Stereogenic Centers	2	2 hours	

4.	Stereochemistry: Configurational Isomers in Cyclic Compounds.	2	2 hours	
5.	Mid-term Exam			
6.	Stereochemistry: Stereochemistry of Organic Reactions.	2	2 hours	
7.	Organic Reactions: Reactivity of Covalent Bonds and Nucleophilic Substitution at Saturated Carbon	2	2 hours	
8.	Organic Reactions: Elimination reactions	2	2 hours	
9.	Organic Reactions: Free Radical Reactions and Addition reactions.	2	2 hours	
10.	Atoms and bonding.	2	2 hours	
11.	Electronic displacements factors.	2	2 hours	
12.	Intermolecular forces. Stability of carbon intermediate, acidity and basicity concepts.	2	2 hours	
13.	Final written & oral			
	Practical topics			
Week No	Topics	No. of hours	Lecture credit hours	Practical credit hours
1.	Laboratory safety rules and Practices	1		1 hour
2.	Identification of organic compounds	1		1 hour
3.	Physical characters (Solubility)	1		1 hour
4.	Chemical Properties (soda lime)	1		1 hour
6.	Mid-term Exam			
7.	Sodium carbonate	1		1 hour
8.	Ferric chloride	1		1 hour
9.	Element test	1		1 hour
10.	Stereochemistry	1		1 hour
11.	Revision	1		1 hour
12.	Practical exam	1		1 hour



5. Teaching and learning Methods:

5.1	Lectures using whiteboard
5.2	Lectures using Data show, PowerPoint presentations
5.3	Research assignments
5.4	Use of computer software for drawing of chemical structures
5.5	Chemical models and Chembio-office Chemdraw
5.6	Animation files

6. Student Assessment:

a- Assessment methods

1. Written exam	To assess understanding, intellectual and professional skills
2. Practical exam	To assess professional and practical skills
3. Oral	To assess knowledge, understanding, intellectual skills, general skills and confidence
4. Laboratory reports	To assess the skills of problem-solving and data presentation

b- Assessment schedule

Assessment 1	Practical	8 th week and 12 th week
Assessment 2	Mid-term	6 th week
Assessment 3	Oral	15 th week
Assessment 4	Written	15 th week

c-Weighting of assessments

1.	Mid-term examination	10 %
2.	Final-term examination	50 %
3.	Oral examination	15 %
4.	Practical examination and Semester work	25 %
Total		100 %

7. List of References

No	Reference	Type
1.	Fundamentals of Organic Chemistry, T. Solomon.	Book



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2.	Notes in organic Chemistry, by Staff members of Pharm. Org. Chem. Dept	Course notes
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8. Matrix of knowledge and skills of the course

No	Course contents	Study Week	ILOS			
			Knowledge & understanding	Intellectual skills	Professional and practical skills	General & transferable skills
1.	Atoms and bonding Electronic displacements factors	1 st	a1	b1	c1	d1
2.	Intermolecular forces Stability of carbon intermediate, acidity and basicity concepts	3 rd	a1	b1	c1	d1
3.	Stereochemistry: Isomerism and Conformational Isomers Stereochemistry: Configurational Isomers	4 th and 5 th	a2	b2	-----	d1
4.	Stereochemistry: Chirality in Systems lacking Stereogenic Centers Stereochemistry: Configurational Isomers in Cyclic Compounds	7 th -9 th	a3	b3	c2	d1
5.	Stereochemistry: Stereochemistry of Organic Reactions. Organic Reactions: Reactivity of Covalent Bonds and Nucleophilic Substitution at Saturated Carbon.	10 th -13 th	a3	b3	c2	d1



Course specification
2019- 2020
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Mansoura University



6.	Organic Reactions: Elimination r Organic Reactions: Free Radical Reactions and Addition reactions	2 nd -5 th and 7 th -12 th	a1, a4	b1,b4	c1, c3, c4	d1, d2, d3, d4
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