



Answer the following questions

[15] Marks

[1]a- For the reaction: $A + \frac{1}{2} B = \frac{3}{2} C + 47.5 \text{ KJ}$; $K_c = 4.3 \times 10^{-5}$

What will be?

- (i) The value of ΔH for the reverse reaction. (3 marks)
- (ii) The value of K_p of the reverse reaction. (3 marks)
- (iii) The influence of pressure on this equilibrium. (3 marks)

b- A solution made by dissolving 4.0g of a solute in 50 g of benzene, freezes at 3.74 °C. If the freezing point of the pure benzene is 5.48 °C. Calculate the molecular weight of the solute ($K_f = 5.12 \text{ }^\circ\text{C}$). (6 marks)

[15] Marks

[2]a- Prove that: $K_p = K_c (RT)^{\Delta n}$. (6 marks)

b- State which of the following statement is correct and which is wrong. Then correct the wrong one.

- (i) At STP, NH_3 will effuse (diffuse) faster than H_2 . (3marks)
- (ii) For oxalic acid ($\text{C}_2\text{O}_4\text{H}_2$), the K_{a2} is greater than K_{a1} . (3marks)
- (iii) The vapour pressure of solution will be less than expected from Raoult's law, if the attraction between different molecules is higher than between similar one. (3 marks)

[15] Marks

[3]a- Give a full account on:
(i) Graham's law of effusion (3marks) (ii) Hess's Law (3 marks)
(iii) Bond energy (3 marks)

b- One mole of CO_2 occupies a volume of 1.32L at 48 °C. Experimentally its pressure was found to be 18.4 atm. Illustrate by the calculation which P will be quite close to 18.4 atm. Either by general gas equation or by applying Van-der Waals equation; knowing that $a = 3.6 \text{ l}^2\text{atm.mol}^{-2}$ and $b = 0.043 \text{ l atm}^{-1}$. (6 marks)

[15] Marks

[4] a- Write briefly on:
(i) Buffer solution (3 marks) (ii) K_w (2 marks)
(ii) Van Hof's factor (2 marks)

b- Choose the correct answer:

- (i) For a mixture of two gases composed of 50% each. (2 marks)
(1) $X_1 = X_2$ (2) $X_1/X_2 = 1$ (3) $X_1 + X_2 = 1$ (4) All of them.
- (ii) When one mole of H_2 gas has been cooled under very high pressure. (2 marks)
(1) The H_2 gas will be liquefied. (2) The H_2 gas will behave similar to ideal gas.
(3) The molecular speed of H_2 molecules will be increase.
- (iii) The pH of a 0.49 M acid solution whose $K_a = 4.53 \times 10^{-9}$ will be (2 marks)
(1) 4.33 (2) 5.22 (3) 4.02
- (iv) Which of the following items when dissolved in water has the highest concentration of magnesium ion? (2marks)
(1) MgF_2 ($K_{sp} = 6.5 \times 10^{-9}$) (2) MgCO_3 ($K_{sp} = 1.8 \times 10^{-11}$)
(3) $\text{Mg}_3(\text{PO}_4)_2$ ($K_{sp} = 1 \times 10^{-25}$)

Best wishes

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امتحان دور مايو 2009م
الفرقة الأولى - المستوى الأول: برامج*
الزمن: ساعتان - التاريخ: 2009/5/27
الدرجة الكلية: 80 درجة



جامعة المنصورة
كلية العلوم - قسم الرياضيات
المادة: رياضيات أساسية
تفاضل وتكامل (112)

*برامج: كيمياء - وكيمياء ونبات - ميكروبيولوجي - كيمياء حيوي - جيوفيزياء - جيولوجيا - كيمياء وحيوان - علوم البيئة

أجب عن الأسئلة الآتية

السؤال الأول: (20 درجة - 5 درجات لكل جزء)

$$f(x) = \sqrt{4-x^2}$$

(أ) أوجد مجال تعريف ومدى الدالة

$$f(x) = \frac{3}{2x-5}$$

(ب) أوجد معكوس الدالة

$$\lim_{x \rightarrow 1} \left[\frac{2}{1-x^2} - \frac{1}{1-x} \right]$$

(ج) أوجد النهاية

$$\lim_{x \rightarrow 0} \frac{3^x - 1}{x}$$

(د) أوجد النهاية

السؤال الثاني: (20 درجة)

[6]

$$y = \frac{(1+x)^5 \sqrt{x^3+2}}{(x-1)^3(x^2+1)}$$

(أ) أوجد $\frac{dy}{dx}$ ، إذا كانت

[6]

$$f(x) = \begin{cases} x^2 - 4 & x \neq 2 \\ x - 2 & \\ A & x = 2 \end{cases}$$

(ب) أوجد قيمة الثابت A ، بحيث تكون الدالة

[8]

(ج) أوجد معادلتى المماس والعمودي للمنحنى $y = f(x) = \sqrt{2x+1}$ عند النقطة $(4,3)$.

السؤال الثالث: (20 درجة - 5 درجات لكل جزء):

أوجد المشتقة الأولى $\frac{dy}{dx}$ لكل من الدوال الآتية:

$$\cos(xy) = y^2 + x \quad (\text{ب})$$

$$y = \text{sech}(\cos^{-1} 2x) \quad (\text{أ})$$

$$y = x^{\sec x} \quad (\text{ع})$$

$$y = 2 \ln(\cot t), \quad x = \tan t + t^3 \quad (\text{ج})$$

السؤال الرابع: (20 درجة - 5 درجات لكل جزء):

احسب التكاملات الآتية:

$$\int \frac{\sqrt{9-x^2}}{x^2} dx \quad (\text{ب})$$

$$\int_1^2 \frac{x^3 - 3x^2 + 1}{\sqrt{x}} dx \quad (\text{أ})$$

$$\int \frac{2x-8}{x^2-3x} dx \quad (\text{د})$$

$$\int x e^{5x} dx \quad (\text{ج})$$



Answer All the questions with labelled diagram

Question One: (15 degree)

Write about the general characters of phylum nematode, and describe the life cycle of two examples of this phylum.

Question Two: (15 degree)

Write briefly on General characters, classification and economic importance of annelida, arthropoda, mollusca and echinodermata.

Question Three: (15 degree)

Write short notes on three only of the following:

- A- General characters of protozoa.
- B- Nutrition and locomotion of *Euglena viridis*.
- C- Nutrition and reproduction in sponges.
- D- Reproduction and life cycle of *Entamoeba histolytica*.

Question Four: Answer three only of the following: (15 degree)

- A- With labeled diagrams only illustrate the following: Metacercaria of *Fasciola* – Sea anemone- Cercaria of *Schistosoma*- Egg of *Ascaris*- Hexacanth of *Taenia* and isolated polyp of *Alcyonium*.
- B- How can you diagnose the infections of *Fasciola*, *Schistosoma*, *Taenia*, *Ascaris* and *Ancylostoma*.
- C- Compare between the medusa of *Obelia* and that of *Aurelia*.
- D- Give an account on life cycle of *Fasciola* or *Taenia* or *Schistosoma*.

With best wishes of success,

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