

Faculty of Science
Zoology Dept.
Level 4th Zoology &
chemistry

Final Exam
Time: 2 hours
Degree : 60
Experimental Embryology

Answer the following questions:

Q1: Write short accounts on the followings:

- 1. Regeneration of lens in amphibian.*
- 2. Ovulation.*
- 3. Role of stem cells in repair & cancer.*
- 4. Functions of Sertoli cells.*

Q2: Illustrate the followings:

- 1. What do you mean by the following definitions: Adult & embryonic stem cells, Plasticity-Antigen, Regeneration.*
- 2. Regeneration of limb.*
- 3. Role of infections and metabolic diseases in the induction of congenital abnormalities.*
- 4. Structural chromosomal abnormalities.*

Q3: Discuss briefly the followings :

- 1. Skull bone sutures and skull abnormalities.*
- 2. amniocytosis.*
- 3. Limb abnormalities.*
- 4. Meningocele & meningoencephalocele.*
- 5. Spina bifida occulata.*

With my best wishes:

Prof. Dr. Hassan Ibrahim El-Sayyad Dr. Heba Atef El-Ghawet

Mansoura University
Faculty of Science
Zoology Department
Subject: Comparative anatomy
Full Mark : 60

First Term
Fourth level: Chemistry & Zoology
Date: 31 December 2011
Time Allowed: 2 hours

Answer the following questions

Q1- Fill in spaces with correct words: (20 Mark half for each space)

- a- In the first tetrapods, the mid-dorsal bones presented by ----- and ----- while the occipital bones presented by ----- and ----- but the otic bones presented only by -----.
- b- The morphological colour change is a slow process related to -----, ----- and ----- and controlled by ----- and endocrine glands including -----, and -----.
- c- In shizocoely, the embryonic mesodermal somites differentiated into -----, ----- and ----- whereas myocoel formed in the upper one while ----- and ----- formed in the other two which give rise in the adult ----- and ----- respectively.
- d- The obstacles of following up the development of body cavity are ----- of the coelom and -----, ----- and ----- of the mesentery as well as the appearance of -----.
- e- The second splanchnic arch formed of -----, ----- and ----- in which the first piece either play a part in ----- as in all ----- or converted to give rise ----- as in all ----- which perform a ----- function.
- f- The top surface of the cheek teeth named ----- which differentiated into -----, -----, ----- and ----- as in -----, -----, ----- and -----.

Q2- Find out the name of the following items and mention two different functions (20 Mark 2 for each)

- a- Sheets of serous membrane extend across the body cavity.
- b- The most familiar integumentary glands characteristic to the majority of aquatic organisms.
- c- A collection of mesenchymal cells formed during the development of any hard structure.
- d- An integumentary structure arise from the modification of multicellular glands in some fishes.

- e- A layer of an epithelial tissues without innervation or vasculature.
- f- Hard structures derived from both layers of skin which associated with the appearance of splanchnocranium.
- g- The most superficial and the hardest hard tissue.
- h- A hard structure of B keratin appear firstly in the first amniotes.
- i- Two different canals belonging to the axial endoskeleton.
- j- A star- like cells represented the second element of the largest system of the body.

Q3 Answer as mentioned between brackets (20 Mark)

- a- Hard tissue contain 25% organic compounds (give the name and origin).
- b- A partition of coelomic cavity specific of some male mammalian species (give the name and one function).
- c- Dermal hard structure characteristic to some males of the first hot-blooded animals (give name and one function).
- d- The splanchnic arches supported the pharyngeal region (give the name their number and the number of the pieces formed each one)
- e- The transverse septum of mammals (give name, its nature and one function).
- f- Hard structures belonging to both exo and endoskeleton (give the names of two different structures, their endoskeletal origin and one function for each.
- g- An epidermal hard structure characteristic to mammals only (give name, component, two different types).

With Best wishes

Prof.Dr. Zeinab El Gohary

Mansoura University
Faculty of Science
Chemistry Department
Course: Physical Chemistry
Date : Jan. 2012



Final Examination
Subject: Chemistry (445)
Fourth level
Full Mark : 60 Marks
Time Allowed : 2 hours

Section (A) Chemical Spectroscopy (30 Marks) الامتحان في صفتين

- 1.a) Write on details on the different types of molecules according to their pure rotation spectra.(7.5 Marks)
b) Explain: Reduced mass –angular momentum for rotational spectra and force constant . (7.5 Marks)
2. a) Explain the vibrational spectra of water molecules. (7.5 Marks)
b) The microwave spectrum of HCl shows a series of lines separated by 2.828 cm^{-1} . Calculate the moment of inertia and the internuclear distance in the molecule. (7.5 Marks)
- ($h=6.62 \times 10^{-27} \text{ erg} \cdot \text{s}$, $N_A = 6.02 \times 10^{23}$, atomic weights : H= 1, Cl = 35.5).

Section (B) Answer the Following Questions: (30 Marks)

I- Choose the response answer: (10 mrks)

- 1- Chemical adsorption is
a) exothermic b) irreversible c) favored by high temperature d) all of these
- 2-The extent of physisorption
a) decrease with rise in temperature b) increases with rise in temperature
c) independent of temperature d) first increases and then decreases with rise in temperature
- 3- The curve indicating the variation of adsorption with temperature at constant pressures known as
a) adsorption isotherm b) adsorption isobar c) adsorption isostere
- 4- When there are no external forces, the shape of a liquid drop is determined by
(a) Surface tension of the liquid (b) Density of liquid (c) Viscosity of liquid (d) Temperature of air only
- 5- The rise of a liquid in a capillary tube does not depend upon
(a) Angle of contact (b) Density of the liquid (c) Radius of the capillary tube (d) Atmospheric pressure
- 6- At critical temperature, the surface tension of a liquid
(a) Is zero (b) Is infinity (c) Is the same as that at any other temperature (d) Can not be determined
- 7- NaCl dissolved (added) in to water than it surface tension is
(a) Decreases (b) Increases (c) Remains same (d) All of these
- 8- Excess pressure inside a soap bubble is
(a) Inversely proportional to its radius (b) Directly proportional to its radius
(c) Directly proportional to square roots of its radius (d) Independent of its radius

9-Point out the correct statement

- (a) Freundlich equation is valid over a limited range of pressure. (b) The constants K and n vary with temp.
(c) Freundlich adsorption equation is purely empirical formula (d) All of the above

10- Surface tension may be defined as

- (a) The work done per unit area in increasing the surface area of a liquid under isothermal condition
(b) The work done per unit area in increasing the surface area of a liquid under adiabatic condition
(c) The work done per unit area in increasing the surface area of a liquid under both isothermal and adiabatic conditions. (d) Free surface energy per unit volume

II Answer the following questions: (20 marks)

1- Write on the following:-1

- a- The Ring method used for measuring surface tension .
b- The Kelvin equation and Kelvin effect.

2 why is surface tension of water greater than surface tension of oil.

3- Explain why water rises while mercury falls in a capillary tube.

4- Determine the surface tension of ethyl alcohol if The density of ethyl alcohol is 0.789 g/cm^3 and values of atomic parachor are of O=19.8, C=9.0, H=15.5

5- The adsorption of a gas on a solid surface was found to follow a Langmuir isotherm with $b = 3.76 \text{ kPa}^{-1}$ at a temperature of 25°C . Determine the pressure of gas required to achieve a fractional surface coverage of 10%.

Examiners : Prof. Dr. Esam Gomaa and Sohier Abd El- Hakam



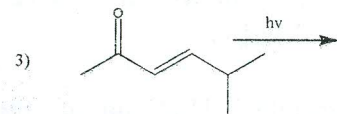
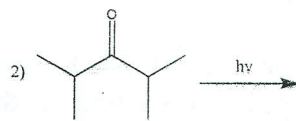
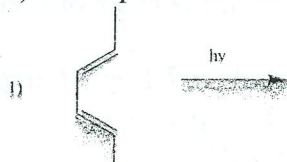
Answer the following questions

1.

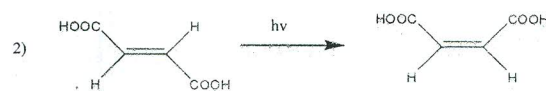
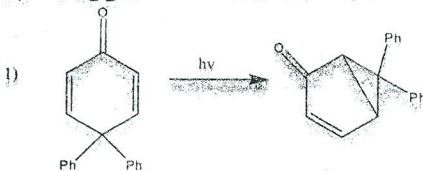
- Write brief account on Jablonski diagram. (7.5 Marks)
- Write short notes on Norrish type II for the photoreaction of carbonyl compounds and explain your answer by an example. (7.5 Marks)

2.

- Complete the following photochemical equations. (7.5 Marks)

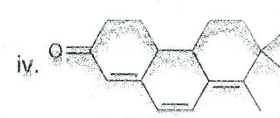
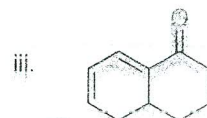
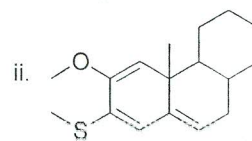
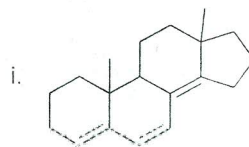


- Suggest the suitable mechanism for the following equations. (7.5 Marks)

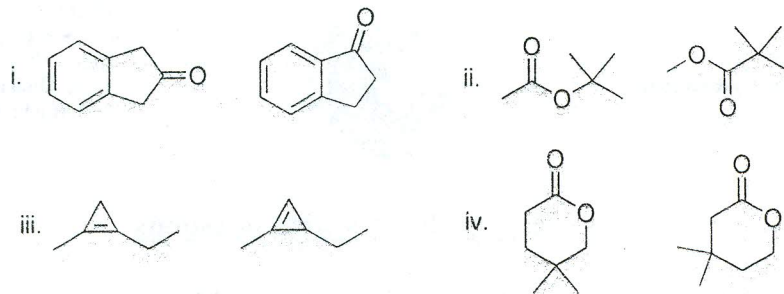


3.

- Calculate λ_{\max} for each of the following (5 Marks):



- How would you use $^1\text{H NMR}$ spectroscopy to distinguish between the following compounds (5 Marks):



c) Illustrate with examples the observed fragmentation pattern in mass spectrometry of (5 Marks):

i. Alcohols.

ii. Ketones.

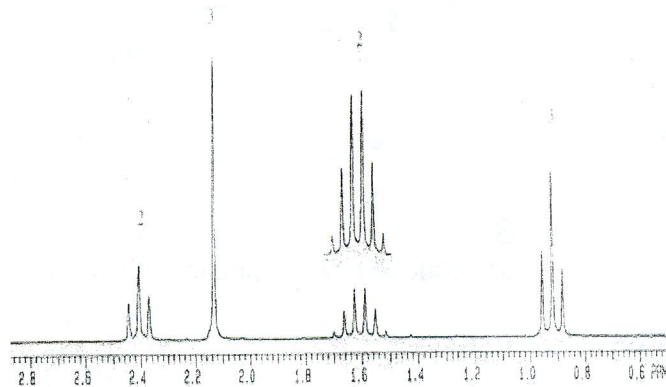
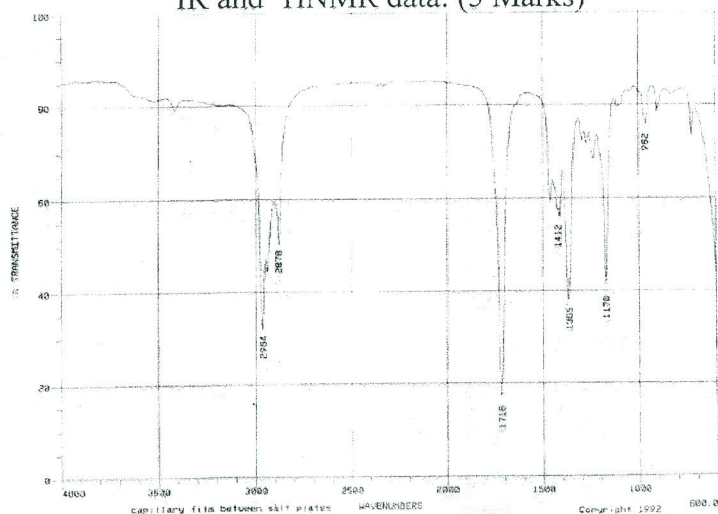
4.

a) Deduce the structure of each of the following compounds on the basis of their ^1H NMR spectra and molecular formulas (5 Marks):

- C_8H_{10} ; δ 1.2 ppm (triplet, 3H), δ 2.6 ppm (quartet, 2H), δ 7.1 ppm (singlet, 5H).
- $\text{C}_{10}\text{H}_{14}$; δ 1.3 ppm (singlet, 9H), δ 7.0 to 7.5 ppm (multiplet, 5H).
- C_6H_{14} ; δ 0.8 ppm (doublet, 12H), δ 1.4 ppm (heptet, 2H).
- $\text{C}_4\text{H}_6\text{Cl}_4$; δ 3.9 ppm (doublet, 4H), δ 4.6 ppm (triplet, 2H).

b) A compound with molecular formula $\text{C}_{10}\text{H}_{10}\text{O}_4$ produces a ^1H NMR spectrum that exhibits only two signals, both singlets. One signal appears at 3.9 ppm with a relative integration value of 79. The other signal appears at 8.1 ppm with a relative integration value of 52. Identify the structure of this compound. (5 Marks)

c) Predict the structure of a compound of molecular formula $\text{C}_5\text{H}_{10}\text{O}$ having the following IR and ^1H NMR data: (5 Marks)



Examiner

Prof. Dr. M. Abo-Elzatab
Dr. Saad Elden Elaraby