

جامعة : المنصورة

كلية : العلوم

قسم / الكيمياء

توصيف مقرر دراسي

١- بيانات المقرر		
المستوى : الثانى	أسم المقرر: Fundamentals of Analytical Chemistry	الرمز الكود: 211 ك
عدد الوحدات الدراسية: ٣ نظرى : ٢ تمارين: - عملى: ٣		البرنامج : الكيمياء

<b>For students undertaking this course, the aims are to:</b>  1 - Introduce the basic information of statistics of experimental data  2 - Introduce the basic information about volumetric, gravimetric, precipitation and compleximetric methods of analysis  3 - Study the general methods for determination of concentration of unknown solution acid, base, halids and metal ions.  4 - Acquire the student the skills of analysis of different types of samples including environmental, chemical and industrial types.	٢- هدف المقرر:
٣- المستهدف من التدريس المقرر:	
<b>a- Knowledge and Understanding :</b>  a - 1 - define the basic concepts of analytical chemistry and statistics of experimental data.  a - 2 - define the basic concepts of various types of titrimetry.  a - 3 - recognize different types of indicators for each type of titrimetry  a - 4 - recognize the application of real samples, analysis and calculations.  <b>On completing this course, students will be able to:</b>	أ-المعلومات والمفاهيم:
<b>b- Intellectual Skills: On completing this course, students will be able to:</b>	ب-المهارات الذهنية

<p>b - 1 - distinguish different types of titrimetric analysis</p> <p>b - 2 - outline suitable methods of various types of chemical analysis.</p> <p>b - 3 - make use of chemical reaction results and calculations</p> <p>b - 4 - choose the suitable indicator from titration curves</p>	
<p><b>c-Professional and Practical Skills: On completing this course,</b></p> <p>c - 1 - assemble glass ware and its uses.</p> <p>c - 2 - carry out a preparation procedure and solutions including preparation of standard solution</p> <p>c - 3 - uses of masking and demasking agents in the analytical procedure</p> <p><b>students will be able to:</b></p>	<p>ج- المهارات المهنية الخاصة بالمقرر:</p>
<p><b>d-General and Transferable Skills: On completing this course,</b></p> <p>d - 1 - work effectively in both team and independently on solving analytical chemistry problems</p> <p>d - 2 - use of IT and search for information</p> <p>d - 3 - communicate effectively with teacher and colleagues.</p> <p><b>students will be able to:</b></p>	<p>د- المهارات العامة :</p>
<p>1- Statistical evaluation of analytical results</p> <p>2- Method of expressing concentration of solution</p> <p>3- Sampling and sample preparation before chemical analysis</p> <p>4- Introduction to titrimetric analysis, stoichiometry of titration reaction and standardization</p> <p>5- Application of titrimetric analysis, acid base titrations and titration curves.</p> <p>6- Theory of indicators in titrimetry</p> <p>7- Oxidation - reduction titrations</p>	<p>٤- محتوى المقرر:</p>

<p>8- Precipitation titrations, methods of detection the end point</p> <p>9- Theory of compleximetry ,application</p> <p>10- Practical</p>																
<p>1 - lectures using data show and board.</p> <p>2 - home works, reports and discussion groups</p> <p>3 - Laboratory work.</p>	<p>٥- أساليب التعليم والتعلم:</p>															
<p><b>The same as normal students, only skeletal disabilities are allowed in the Faculty of Science.</b></p>	<p>٦- أساليب التعليم والتعلم للطلاب ذوي القدرات المحدودة:</p>															
<p>٧- تقويم الطلاب :</p>																
<p><b>7- Student Assessment Methods</b></p> <table border="1" data-bbox="197 1021 936 1608"> <tr> <td><b>Practical exam</b></td> <td><b>To assess</b></td> <td>b1-b4,c1-c3</td> </tr> <tr> <td><b>Final exam</b></td> <td><b>to assess</b></td> <td>a1-a4, b1-b4</td> </tr> <tr> <td><b>Oral exam</b></td> <td><b>to assess</b></td> <td>a1-a4, d3</td> </tr> <tr> <td><b>Mid-term exam</b></td> <td><b>To assess</b></td> <td>a1-a4, b1-b4</td> </tr> <tr> <td><b>Report</b></td> <td><b>to assess</b></td> <td>d1-d3</td> </tr> </table>	<b>Practical exam</b>	<b>To assess</b>	b1-b4,c1-c3	<b>Final exam</b>	<b>to assess</b>	a1-a4, b1-b4	<b>Oral exam</b>	<b>to assess</b>	a1-a4, d3	<b>Mid-term exam</b>	<b>To assess</b>	a1-a4, b1-b4	<b>Report</b>	<b>to assess</b>	d1-d3	<p>أ- الأساليب المستخدمة :</p>
<b>Practical exam</b>	<b>To assess</b>	b1-b4,c1-c3														
<b>Final exam</b>	<b>to assess</b>	a1-a4, b1-b4														
<b>Oral exam</b>	<b>to assess</b>	a1-a4, d3														
<b>Mid-term exam</b>	<b>To assess</b>	a1-a4, b1-b4														
<b>Report</b>	<b>to assess</b>	d1-d3														
<p><b>Assessment Schedule</b></p> <table border="1" data-bbox="197 1686 1168 1980"> <tr> <td><b>Assessment 1</b></td> <td><b>Week #final exam</b></td> <td><b>Week 14</b></td> </tr> <tr> <td><b>Assessment 2</b></td> <td><b>Week #oral exam</b></td> <td><b>Week 14</b></td> </tr> <tr> <td><b>Assessment 3</b></td> <td><b>Week #practical exam</b></td> <td><b>Week 12</b></td> </tr> </table>		<b>Assessment 1</b>	<b>Week #final exam</b>	<b>Week 14</b>	<b>Assessment 2</b>	<b>Week #oral exam</b>	<b>Week 14</b>	<b>Assessment 3</b>	<b>Week #practical exam</b>	<b>Week 12</b>	<p>ب- التوقيت :</p>					
<b>Assessment 1</b>	<b>Week #final exam</b>	<b>Week 14</b>														
<b>Assessment 2</b>	<b>Week #oral exam</b>	<b>Week 14</b>														
<b>Assessment 3</b>	<b>Week #practical exam</b>	<b>Week 12</b>														

Assessment 4	Week #mid-term exam	Week 4,8,12		
Assessment 5	Week #report	Week 10		
<i>Weighting of Assessments</i>			ج- توزيع الدرجات :	
Final-Term Examination	60%			
Oral Examination	10%			
Practical Examination	20%			
Semester work	0%			
Mid-term examination	10%			
Other types of assessment	0%			
Total	100%			
٨- قائمة الكتب الدراسية والمراجع :				
1 - fundamentals of analytical chemistry			أ- مذكرات:	
			ب- كتب ملزمة	
1 - Vogel's - Textbook of Quantitative Chemical Analysis (5th Edition)			ج- كتب مقترحة :	
			د- دوريات علمية أو نشرات..	

مصنوفة المعارف والمهارات المستهدفة من المقرر الدراسي

	المحتويات للمقرر	أسبوع الدراسة	المعارف الرئيسية	مهارات ذهنية	مهارات مهنية	مهارات عامة
1	Statistical evaluation of analytical results	1-2	a1	b3		d2
2	Method of expressing concentration of solution	3-4	a1	b1,b2	c1,c2	d2,d3

3	Sampling and sample preparation before chemical analysis	5-6	a1	b1,b2	c2	d1,d2,d3
4	Introduction to titremetric analysis,stoichiometry of titration reaction and standardization	7-8	a2	b1,b2		d2
5	Application of titremetric analysis, acid base titrations and titration curves.	9	a4	b1,b2,b 3	c2	d3
6	Theory of indicators in titrimetry	10	a3	b1,b2,b 3		d2
7	Oxidation - reduction titrations	11-12	a4	b2,b3	c2	d2
8	Precipitation titrations, methods of detection the end point	13	a2,a4	b2,b3	c2	d2
9	Theory of compleximetry ,application	14	a2,a4	b2,b3	c2	d2,d3
10	Practical	1-12			c1,c3	d1,d3

أستاذ المادة : ا.د/ محمد محمد محمد الدفراوى

رئيس مجلس القسم العلمي : أ.د /سالم السيد سمرة